INTRODUCTION

At present the developed countries have accepted a concept of the best available technologies, which means modernization of all the production with the purpose of minimizing a negative effect on environment via maximal processing of raw materials and by-products of production, reducing expenses on reagents and water, ensuring the possibility of water recirculation at enterprises. Membrane processes were proven to be successful in solving these tasks [1].

The introduction of membrane technologies at milk-processing enterprises allows enhancing the efficiency and cost effectiveness due to the economy of energy resources, more complete use of raw material resources, expanding the assortment, receiving additional profit [2].

Among modern membrane technologies, including reverse osmosis, microfiltration, ultrafiltration, nanofiltration, and electrodialysis, in Ukraine ultrafiltration, nanofiltration and electrodialysis found their practical application.

During nanofiltration (NF) there is concentration of dry substances up to 18–22 % which makes it rea-
reasonable to use it with the purpose of reducing energy resources compared to whey evaporation in vacuum. Besides, from the practical standpoint the optimal variant of NF-processing is maximal removal of mineral salts and lactic acid from different kinds of milk whey with the most complete retaining of valuable whey components – proteins and lactose, and, as a result, obtaining concentrates, the technological indices of which allow using them in the production of other products [3–6].

Electrodialysis (ED) allows increasing the target indices of demineralization up to 90 %, which is especially promising for processing of salty cheese, acid and caseic milk whey [7, 8]. Any kind of whey with the application of demineralization of different level (50, 70, 90 % and above) may be standardized by physical-chemical composition and organoleptic indices, it is possible to achieve the category of quality which allows using it in baby food [2].

However, achieving a high level of demineralization is accompanied with a considerable increase in energy expenses, which is economically not substantiated [9]. Taking into consideration the fact that usually not more than 70–80 % of salts are practically removed, electrodialysis is widely used in industrial conditions while desalinizing various kinds of milk whey [10–12]. To increase efficiency, electrodialysis is combined with other membrane methods of separation [2, 13]. In particular, the combination of nanofiltration and electrodialysis is recommended not only to enhance the efficiency of whey processing technology, to economize energy resources, but also to reduce the impact of high temperatures on thermolabile components of milk whey which, at the end, enhances the biological value and improves technological properties of obtained products [14]. Such demineralized dry whey has better taste, physical-chemical characteristics and functional-technological properties compared to dry whey, obtained by traditional technology.

Taking the abovementioned into consideration, one may assume that the use of a complex of membrane methods allows increasing the quality of milk whey processing in conditions of a dairy enterprise compared to their separate application. This technology was successfully implemented at some dairy enterprises.

MATERIALS AND METHODS

Cheese and acid whey, obtained during the production of cheese or lactic cheese and the corresponding kinds of whey after nanofiltration and electrodialysis, were used in the work. Demineralization of milk whey was conducted at experiment electrodialysis (MEGA, Czech Republic) and nanofiltration equipment (GEA, Denmark). Dry samples were obtained by drying the corresponding kinds of whey on spray dryer.

The mass content of moisture in dry products was defined by the standardized method, which is based on the ability of the product to lose free moisture while drying at constant temperature – (102 ± 2) °C. The arithmetic mean value of the results of two parallel measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.06 %.

The mass content of ash was determined by the method, based on ashing 2.8–3.2 g of dry product at the temperature of (525 ± 25) °C. The mass content of ash in percentage (X) was calculated according to the formula:

\[
X = \left( \frac{m_1 - m_2}{m} \right) \cdot 100, \tag{1}
\]

where \(m_1\) – mass of a pot with the ashes of the product after ashing, g; \(m_2\) – mass of an empty pot after calcination, g; \(m\) – mass of the weighed quantity of product, g; 100 – coefficient of transferring grams into percentage.

The arithmetic mean value of the results of two parallel measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.1 %.

The mass content of fat was determined by the standardized acid method, based on extracting fat from dry products under the impact of concentrated sulphuric acid and isomalt alcohol with further centrifugation and measuring the volume of fat in the calibrated part of butyrometer. The arithmetic mean value of the results of two parallel measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.5 % on condition that the results were within one lowest graduation mark of the butyrometer.

The mass content of lactose was determined by the standardized iodometric method in the weighed quantity of the product of 3.0 g. The arithmetic mean value of the results of two parallel measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.2 %.

The acidity of dry whey was determined by the standardized titrimetric method using 0.1 mol/cu dm. The arithmetic mean value of the results of two parallel
measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.5 %.

The solubility index was determined in one cubic centimeter by the method, based on measuring the volume of insoluble precipitation in the restored sample of dry whey after centrifugation at 8,000 rpm for 5 min. The arithmetic mean of the results of two parallel measurements of one sample was accepted as the final result, the permissible differences between them could not exceed 0.1 %.

The foam-forming capability of dry samples was determined by the relative increase in their solution volumes after shaking. For this purpose, a chemical glass was added weighed 25 g of dry product, and 225 g of distilled water with the temperature of 20 °C. The samples of 250 cu cm were shaken for 5 min at the frequency of shaker rotation of 800 rpm. After shaking the volume of liquid fraction and the volume of obtained foam were measured by a measuring glass cylinder. The foam-forming capability \((C, \%)\) was determined by the formula:

\[
C = \frac{V_f}{V_m} \cdot 100,
\]

where, \(V_f\) – volume of foam after shaking, cc; \(V_m\) – the initial volume of the mixture prior to shaking, cc; 100 – coefficient of transferring into percentage.

The arithmetic mean value of the results of two parallel measurements was accepted as the final result after rounding down to the first decimal figure.

The moisture-retaining capability of dry products was determined by the increase in the mass of wet precipitate after centrifugation. A previously weighed centrifugal tube was introduced a weighed quantity in the amount of 1 g and 3 cc of distilled water. The mixture was mixed for 1 min. Then the tube was centrifuged at 8,000 rpm for 15 min. The liquid, which was above the precipitate, was poured out, the tube was turned over the filtration paper and left undisturbed for 10 min (to remove the remaining refined oil) and weighed.

The moisture-retaining capability \((MRC, \%)\) was calculated by the formula:

\[
MRC = \frac{C-B}{B-A} \cdot 100,
\]

where, \(A\) – mass of an empty centrifugal tube, g; \(B\) – mass of centrifugal tube with the weighed quantity of dry matter, g; \(C\) – mass of centrifugal tube with precipitation after centrifugation, g.

The fat-retaining capability of the investigated products was estimated using the emulsion solutions with refined oil. A previously weighed centrifugal tube was introduced a weighed quantity of dry product in the amount of 1 g and 3 cc of refined oil. The mixture in the tube was mixed for 1 min. Then the tube was centrifuged at 8,000 rpm for 15 min. The liquid, which was above the precipitate, was poured out, the tube was turned over the filtration paper and left undisturbed for 10 min (to remove the remaining refined oil) and weighed.

The fat-retaining capability \((FRC, \%)\) was calculated by the formula:

\[
FRC = \frac{C-B}{B-A} \cdot 100,
\]

where, \(A\) – mass of an empty centrifugal tube, g; \(B\) – mass of centrifugal tube with the weighed quantity of dry matter, g; \(C\) – mass of centrifugal tube with precipitation after centrifugation, g.

The emulsifying capability of the investigated products was estimated using the emulsion solutions with refined oil. The chemical glass with the volume of 500 cc was introduced 7 g of dry product and 100 cc of distilled water. The mixture was mixed using the mixer at 4,000 rpm for 5 min with subsequent addition of 100 cc of refined oil and the mixing was continued at 8,000 rpm for 5 min. The emulsion was poured in equal parts into 4 calibrated centrifugal tubes with the volume of 10 cc and centrifuged at 2,000 rpm for 5 min. The emulsifying capability \((EC, \%)\) was calculated by the formula:

\[
EC = \frac{V}{V_1} \cdot 100,
\]

where, \(V\) – volume of the liquid above the precipitate, cc; \(V_1\) – total volume of centrifugal tube (10 cc); 100 – coefficient of transferring into percentage.

The arithmetic mean value of the results of two parallel measurements was accepted as the final result after rounding down to the first decimal figure.

The mathematical processing of the results was conducted by methods of statistical analysis and standard algorithms of Microsoft Excel programs. The experiments were conducted in three repeats. The results were deemed to be reliable at \(P < 0.05\).
RESULTS AND DISCUSSION

Our previous studies established that during electrodialysis the mass share of ash in the initial whey decreased in the range from 0.56–0.71 % to 0.02–0.08 % after electrodialysis, depending on the kind of whey and the initial content of ash therein. The maximal decrease in the content of mineral salts in cheese milk whey was achieved using nanofiltration at the level of 40 % [9, 10]. Regardless of different levels of demineralization, there was the most considerable decrease noted in the content of monovalent ions which led to improving organoleptic properties of dry whey [10]. This whey may be considered to be full value raw material during the production of other food products – cooked sausages, yogurts, ice-cream, cheese paste, cheeses, etc.

Taking the abovementioned into consideration, there was a study of the impact of combined application of membrane methods of processing whey on the composition and technological properties of the end products. Being the most common by-products of milk processing, formed during the production of cheese and sour-milk cheese, cheese and acid milk whey are usually processed by drying. Thus, the most attractive and economically grounded method is a possibility of improving the consumer properties of such dry products due to a high content of complete whey proteins therein.

The organoleptic and physical-chemical indices of liquid and dry products of processing cheese and acid whey were determined. It is noteworthy that in addition to dry kinds of whey, liquid demineralized whey with the mass share of dry substances of ≈ 20 % is of some interest for practical application, for instance, for the production of sour-milk beverages. As noted above, the decrease in the content of ash in whey improves its taste properties considerably. The data, presented in Table 1, demonstrate that after electrodialysis the indices of the mass share of ash (1) decreased in cheese and acid whey by 21.2 and 62.7 % and after the treatment using both methods – by 9.6 and 14.7 % respectively, compared to the initial content in the initial whey. The same tendency was remarked regarding the acidity indices as well: the values decreased for cheese and acid whey (after electrodialysis) 1.8 times and 4.2 times and 1.2 times and 2.7 times respectively, after the combination of treatment methods. Therefore, the decrease in the content of salts and lactic acid leads to improving organoleptic and physical-chemical indices of the end products.

It was established that during the treatment of whey with nanofiltration or during complex treatment with nanofiltration and further electrodialysis, the content of dry substances in liquid concentrate increased to 19...20 %. This intermediate product of whey processing is full value raw material and may be used for normalization of milk mixtures while producing other milk products, sour milk beverages, etc.

It is obvious that the application of any method of processing whey or their combination allows improving the properties of the initial raw material considerably due to decreasing the content of ash (Table 2). For instance, after consecutive treatment using the methods of nanofiltration and electrodialysis, the indices of ash in dry whey (NF/ED) decreased 2.8 times in case of using cheese whey as the initial raw material and 3 times – in case of acid whey. The demineralization level during electrodialysis may reach 86.5 % for cheese whey and 95.8 % for acid whey, and during the treatment using the combination of methods – up to 90 % and 75 %.

To estimate the possibility of using dry demineralized whey (NF/ED), there was a determination of its function-

| Table 1. The physical-chemical indices of liquid products of processing cheese and acid whey after different methods of treatment |
|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|
| Index           | Whey initial    | Liquid concentrate after nanofiltration (NF) | Dilute after electrodialysis (ED) | Liquid concentrate combined method of treatment (NF/ED) |
|                 | cheese          | acid            | cheese          | acid            | cheese          | acid            | cheese          | acid            |
| Mass share of dry substances, % | 6.67            | 5.77            | 19.43           | 15.56           | 6.04            | 5.48            | 20.03           | 17.5            |
| Mass share of ash, % | 0.52            | 0.75            | 1.0             | 1.1             | 0.41            | 0.28            | 0.47            | 0.64            |
| Mass share of lactose, % | 4.50            | 4.02            | 15.20           | 10.43           | 4.90            | 5.0             | 6.71            | 5.92            |
| Mass share of fat, % | < 0.1           | < 0.1           | 0.1             | 0.1             | < 0.1           | < 0.1           | 0.1             | 0.1             |
| Titrated acidity, 0T | 14.5            | 75.6            | 10.0            | 120             | 8.0             | 18              | 12.0            | 28              |
al-technological indices – foam-forming (2), moisture-retaining (3), fat-retaining (4) and emulsifying (5) properties (Table 3). These properties characterize the ability of whey proteins to participate in surface phenomena and are most widely used while obtaining products on the basis of foam-like and emulsion systems. It is evident that such differences are possible due to the increase in protein content in dry demineralized whey.

The presented data demonstrate that the highest indices of moisture-retaining and fat-retaining capability were found for cheese whey, obtained by the combination of treatment methods, namely, 32.5 % and 120 % respectively. It is quite evident that it is due to the increased content of protein and the ability of whey proteins to bind water, emulsify and retain fats, dry cheese demineralized whey has better technological properties. A similar regularity was noted for foam-retaining capability as well.

Summarizing the abovementioned, one may assume that dry demineralized whey may be used as full value

Table 2. The characteristics of dry milk whey after treatment with different membrane methods

<table>
<thead>
<tr>
<th>Name of indices</th>
<th>Dry whey (traditional technology)</th>
<th>Dry demineralized whey</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>whey cheese</td>
<td>acid whey after nanofiltration (NF)</td>
</tr>
<tr>
<td>Mass content of dry substances, %</td>
<td>97.00</td>
<td>95.19</td>
</tr>
<tr>
<td>Mass content of ash, %</td>
<td>7.27</td>
<td>8.29</td>
</tr>
<tr>
<td>Mass content of lactose, %</td>
<td>74.5</td>
<td>73.03</td>
</tr>
<tr>
<td>Mass content of fat, %</td>
<td>1.50</td>
<td>1.6</td>
</tr>
<tr>
<td>Mass share of protein, %</td>
<td>12.57</td>
<td>11.10</td>
</tr>
<tr>
<td>Titrated acidity, 0T</td>
<td>14.0</td>
<td>75.0</td>
</tr>
<tr>
<td>Index of solubility cc of wet precipitate</td>
<td>0.3</td>
<td>0.5</td>
</tr>
<tr>
<td>Level of demineralization, %</td>
<td>–</td>
<td>–</td>
</tr>
</tbody>
</table>

Organoleptic indices

<table>
<thead>
<tr>
<th>Consistence</th>
<th>Fine powder</th>
</tr>
</thead>
<tbody>
<tr>
<td>Taste and smell</td>
<td>Sweetish-salty taste</td>
</tr>
<tr>
<td>Color</td>
<td>Light yellow color</td>
</tr>
</tbody>
</table>

Table 3. The functional-technological properties of dry whey

<table>
<thead>
<tr>
<th>Name of product</th>
<th>Foam-forming ability, %</th>
<th>Moisture-retaining ability, %</th>
<th>Fat-retaining ability, %</th>
<th>Emulsifying ability, %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dry cheese whey (control)</td>
<td>5.8 ± 0.6</td>
<td>12.2 ± 0.1</td>
<td>83.0 ± 0.6</td>
<td>27.0 ± 0.4</td>
</tr>
<tr>
<td>Dry acid whey (control)</td>
<td>4.3 ± 0.2</td>
<td>14.6 ± 0.1</td>
<td>79.0 ± 0.2</td>
<td>29.0 ± 0.1</td>
</tr>
<tr>
<td>Dry cheese whey (NF/ED), demineralization level 90 %</td>
<td>15.6 ± 0.2</td>
<td>32.5 ± 0.01</td>
<td>120.0 ± 0.2</td>
<td>33.0 ± 0.2</td>
</tr>
<tr>
<td>Dry acid whey (NF/ED), demineralization level 75 %</td>
<td>11.9 ± 0.1</td>
<td>27.4 ± 0.02</td>
<td>107.3 ± 0.1</td>
<td>31.8 ± 0.1</td>
</tr>
</tbody>
</table>
replacement of dried skimmed milk and dry whey in the formulations of other food products with the purpose of improving their consumer and functional-technological properties.

RESULTS

It was established that there was high efficiency of applying membrane methods for processing of secondary resources of milk raw materials in current conditions of raw materials source, which are presented by milk whey, formed during cheese production. It was determined that processing of different kinds of whey using the combination of nanofiltration and electrodialysis methods led to a considerable decrease in the content of ash compared to the initial whey. The level of demineralization of cheese whey may amount to 90 %, that of acid whey – 75 %. In addition to dry kinds of whey, liquid demineralized whey is of some interest for practical application, which may be used during the production of sour-milk and milk-containing drinks due to a high content of dry substances. It was found that the increase in protein content in dry demineralized whey, obtained using the complex of membrane methods of processing, led to a considerable increase in its foam-forming, moisture-retaining, fat-retaining and emulsifying abilities compared to milk whey, obtained by a traditional technology.

CONCLUSIONS

It was established that dry demineralized whey, obtained by a combination of nanofiltration and electrodialysis methods, had better organoleptic and physical-chemical indices compared to dry whey. The investigated industrial samples are remarkable for improved functional and technological properties which allows using them in the formulations of other food products.
чено высокую эффективность применения мембранных методов для переработки вторичных ресурсов молочного сыра в существующих условиях сырьевой базы, которыми на сегодняшний день является сыроварка молочная, которая получается при производстве сыра. Установлено, что обработка различных видов сырьевой молочной изменение содержания золы по сравнению с исходным сыром. Уровень деминерализации подсывочной сырьевой молочной может достигать 90%, кислоты сыворотки – 75%. Кроме сухих видов сырья молочной, определенную заинтересованность для практического применения имеет жидкость, деминерализованная сыворотка, которая благодаря высокому содержанию сухих веществ может использоваться в производстве кисломолочных и молокосодержащих напитков. Установлено, что с увеличением содержания белка в сухой деминерализованной сыворотке, полученной с помощью комплекса мембран - методов обработки, ее пенообразующая, влаго-удерживающая, жироудерживающая и эмульгирующая способность по сравнению с сырьевой молочной, полученной по традиционной технологии, существенно возрастает. Выводы. Установлено, что сухая деминерализованная сыворотка, полученная с использованием комбинации методов фильтрации и электродиализа, имеет лучшие органолептические и физико-химические показатели по сравнению с сырьевой сухой. Исследованные промышленные образцы характеризуются улучшением функционально-технологическими свойствами, что позволяет использовать их в производстве пищевых продуктов.

Ключевые слова: фильтрация, электродиализ, комбинированные мембранные методы, физико-химические показатели, уровень деминерализации, сырьевой молочный деминерализованный сухой, функционально-технологические свойства.

REFERENCES